



PAP-003-001529

Seat No. _____

B. Sc. (Sem. V) (CBCS) Examination

October / November - 2018

Biochemistry : Paper - 501

(Enzymology)

Faculty Code : 003

Subject Code : 001529

Time : $2\frac{1}{2}$ Hours]

[Total Marks : 70

1 Answer the following questions : 1×20=20

- (1) The site of the enzymic protein which is involved in biochemical reaction is known as?
- (2) The enzyme where catalysis involves the transfer of functional group are named as?
- (3) Define Isoenzymes.
- (4) What is complete enzyme including the prosthetic group is called as?
- (5) Name the model proposed by Koshland.
- (6) Name the coenzymes used by Dehydrogenase.
- (7) Which metal ion out of Na, Fe⁺³, K, Zn, is more important in the enzyme catalysis?
- (8) E. C. 4 : 1 : 2 : 1 is the enzyme no. in which 4 stands for?
- (9) Define specific activity.
- (10) Which technique is used to separate a protein that binds strongly to its substrate?
- (11) To separate enzyme based on the isoelectric pH the technique used is?
- (12) Define Salting out process
- (13) Give effect of competitive inhibitor on Km and Vmax.

- (14) Define First order reaction.
- (15) How will you differentiate whether enzyme is allosteric or Non Allosteric?
- (16) Enzyme Tranaminase follows which type of bisubstrate reaction.rection?
- (17) Name the enzyme raised in first five days is of acute pancreatitis.
- (18) Which enzyme gets elevated in disease of biliary tract ?
- (19) Which fungus is used to prepare blue cheese?
- (20) Define question 10.

2 Answer the following questions :

(a) Give answers of any **three** questions : **2×3=6**

- (1) Define : Ribozyme.
- (2) What do you understand from absolute specificity?
(With example)
- (3) Why one should isolate and purify enzyme?
- (4) Why do we get hyperbola curve of v Vs $[S]$?
- (5) How sweetening agents are prepared from corn starch?
- (6) Write about Dye-ligand chromatography.

(b) Give answers of any **three** questions : **3×3=9**

- (1) Write a note on multi enzyme complex with suitable example.
- (2) Write a note on acid base catalysis.
- (3) Write various methods of enzyme immobilization.
- (4) Giving example explain ordered single displacement reaction.
- (5) Briefly explain use of lipase enzyme for diagnostic purposes.
- (6) Write the differences between chemical and biological catalyst.

(c) Give answers of any **two** questions : **2×5=10**

- (1) Discuss different methods used for enzyme assay.
- (2) Write a detailed note on Lactate dehydrogenase as an example of Isoenzyme.
- (3) Explain the process of cheese making and bread making.
- (4) Derive MM equation of enzyme catalyzed reaction.
- (5) Write a note on different methods used for separation of enzyme on basis of Size or Mass.

3 Answer the following questions :

(a) Give answers of any **three** questions : **2×3=6**

- (1) Define with example: Metalloenzyme.
- (2) Differentiate endopeptidases with exopeptidases with examples.
- (3) Write in brief about biosensor.
- (4) Define K_m and give its significance.
- (5) Write various isoenzyme of creatine kinase and write their uses in diagnosis of various diseases.
- (6) Explain how enzyme deficiency leads to PKU.

(b) Give answers of any **three** questions : **3×3=9**

- (1) Explain chemical as well as enzymatic method for uric acid determination.
- (2) Write a note on cofactors.
- (3) Discuss difference between affinity chromatography and affinity elution.
- (4) State effect of competitive and non competitive inhibition on K_m and V_{max} of reaction.
- (5) How alcoholic beverages are prepared?
- (6) State important characteristics of allosteric enzymes.

(c) Give answers of any **two** questions : **2×5=10**

- (1) Write a detailed note: Role of I.U.B. in the scheme of enzyme classification.
- (2) Briefly describe the enzyme characteristics :
 - (a) Colloidal nature
 - (b) Geometric specificity.
- (3) Describe the role of alkaline phosphatase in the diagnosis of various diseases.
- (4) Giving example explain positive and negative regulation of allosteric enzyme.
- (5) Explain any two methods of enzyme purification which is based on charge.
